

THE SYNTHESIS, PROPERTIES AND STRUCTURES OF SOME UNSATURATED METHYL- α ,D-GLUCOPYRANOSIDE ACETALS

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Abstract—A method for the synthesis of unsaturated methyl- α ,D-glucopyranoside acetals has been developed. The structure and some properties of these acetals and their derivatives has been investigated.

THE acetals formed from unsaturated aldehydes such as citral,¹ cinnamaldehyde,² crotonaldehyde³ and pentaerythritol with simple alcohols are known.

The unsaturated acetals of sugars have not been described and in a previous investigation⁴ only the reaction product of methyl- α ,D-glucopyranoside with crotonaldehyde was described. Sugar derivatives of this type may be given different structural formulae and the present work describes syntheses and the structural determination of unsaturated methyl- α ,D-glucopyranoside acetals with cinnamaldehyde, crotonaldehyde and citral.

The synthesis of unsaturated acetals of sugars, presents difficulties, as in the majority of cases the catalyst react with the unsaturated aldehyde, or cause, polymerization of aldehyde. For this reason acetals could not be prepared from aldehydes and methyl- α ,D-glucopyranoside in the presence of hydrogen chloride, zinc chloride or an ion exchanger but the azeotropic removal of water at elevated temperatures, and the use of anhydrous magnesium sulphate as catalyst, resulted in the desired compounds.

In comparison with the synthesis of saturated acetals, the reaction of methyl- α ,D-glucopyranoside with unsaturated aldehydes, requires longer time, higher temperatures and the azeotropic removal of water. The structure of the reacting aldehyde is also important. The acetal of cinnamaldehyde can be obtained easily, in a 80% yield at 86–89° and with a 1:6 ratio of glucoside to aldehyde, while only a 37% yield of acetal is obtained from crotonaldehyde under similar reaction conditions and the synthesis of citrylidene acetal requires a temperature of 150°. The reactivity of the unsaturated aldehyde is weaker the longer the chain linkage, and stronger when a phenyl radical is present. Analysis of the acetals has shown that they are equimolar compounds of aldehyde and glucoside.

For the purpose of structural determination and resolution of free hydroxyl groups, the mono acetals were subjected to etherification with methyl iodide, with formation of dimethyl ethers which by action of benzaldehyde in presence of ZnCl₂, according to Freudenberg *et al.*⁵ resulted in the formation of 2,3 dimethyl-4,6 benzyldene-methyl- α ,D-glucopyranoside, previously described.

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2,3 Dimethyl-4,6 crotonylidene-methyl- α ,D-glucopyranoside (IV) was obtained from 2 g acetal (I) and 30 ml methyl iodide in the presence of Ag_2O as catalyst in a 62% yield which crystallized as white fine crystals m.p. $46-47^\circ$; (α) $_{\text{D}}^{25}$ - 117.4° (c. 1.0 in CHCl_3). Found: C, 57.0; H, 7.9. $\text{C}_{13}\text{H}_{21}\text{O}_6$ requires: C, 56.9; H, 8.0%.

2,3 *Diacetyl-4,6 crotonylidene-methyl- α ,D-glucopyranoside* (V) and 2,3 *Diacetyl-4,6 cinnamylidene-methyl- α ,D-glucopyranoside* (VI) were the esterification products of 2 g acetal (I, II) with 2.5 ml acetic anhydride in pyridine at room temp, yield 95%. 2,3 *Diacetyl-4,6 crotonylidene-methyl- α ,D-glucopyranoside* is a syrup, $(\alpha)_D^{25} + 102.6^\circ$ (c. 0.5). Found: C, 54.5; H, 6.8; $C_{18}H_{22}O_8$ requires: C, 54.5; H, 6.7%. 2,3 *Diacetyl-4,6 cinnamylidene-methyl- α ,D-glucopyranoside* crystallized from methanol, m.p. 163–164°; $(\alpha)_D^{25} + 78.2^\circ$ (c. 1.8 in $CHCl_3$). Found: C, 61.3; H, 6.2; $C_{20}H_{24}O_8$ requires: C, 61.2; H, 6.2%.

2,3 *Dibenzoyl-4,6-crotonylidene-methyl- α ,D-glucopyranoside* (VII) and 2,3 *Dibenzoyl-4,6 cinnamylidene-methyl- α ,D-glucopyranoside* (VIII) 2 g of acetal (I, II) in 10 ml pyridine were treated with 2.8 ml benzoyl chloride at -10° , yield 50–60%. 2,3 *Dibenzoyl-4,6 crotonylidene-methyl- α ,D-glucopyranoside* crystallized from ethanol m.p. 142–143°, $(\alpha)_D^{25} + 135.8^\circ$ (c. 1.0 $CHCl_3$). Found: C, 65.8; H, 5.6; $C_{26}H_{24}O_8$ requires: C, 66.0; H, 5.7%. 2,3 *Dibenzoyl-4,6 cinnamylidene-methyl- α ,D-glucopyranoside*, crystallized from ethanol, m.p. 174–176°, $(\alpha)_D^{25} + 96.6^\circ$ (c. 1.7 in $CHCl_3$). Found: C, 69.6; H, 5.4; $C_{28}H_{24}O_8$ requires: C, 69.7; H, 5.5%.

2,3 *Di-p-nitrobenzoyl-4,6-citrylidene-methyl- α ,D-glucopyranoside* (IX) was produced from 3 g acetal (III) in pyridine and 12 g *p*-nitrobenzoyl chloride. The yellow crystalline product was recrystallized from ethanol, m.p. 76–78°, $(\alpha)_D^{25} + 132.3^\circ$ (c. 0.6) in $CHCl_3$. Found: C, 59.7; H, 5.2; $C_{31}H_{24}O_{11}N_2$ requires: C, 59.4; H, 5.4%.

2-*p*-Toluenesulphonyl-4,6 *crotonylidene-methyl- α ,D-glucopyranoside* (X) and 2,3 *Di-p-Toluenesulphonyl-4,6 cinnamylidene-methyl- α ,D-glucopyranoside* (XI) were produced from the acetals (I, II) and *p*-toluenesulphonyl chloride in molar ratios of 1:2 and 1:6 respectively and at room temp. 2-*p*-Toluenesulphonyl-4,6 *crotonylidene-methyl- α ,D-glucopyranoside* has a m.p. 49–51°, $(\alpha)_D^{25} + 66.9^\circ$ (c. 1.6). Found: C, 54.3; H, 6.0; S, 6.7; $C_{18}H_{24}O_8S$ requires: C, 54.0; H, 6.0; S, 8.0%. 2,3 *Di-p-Toluenesulphonyl-4,6-cinnamylidene-methyl- α ,D-glucopyranoside*, crystallized from ethanol, m.p. 102–104°, $(\alpha)_D^{25} + 32.2^\circ$ (c. 2.8 in $CHCl_3$). Found: C, 58.4; H, 5.3; $C_{20}H_{24}O_{10}S_2$ requires: C, 58.4; H, 5.2%.

2,3 *Dimethyl-4,6 benzylidene-methyl- α ,D-glucopyranoside* obtained from acetals (I, II) was converted into the dimethyl derivative 4,6-benzylidene-methyl- α ,D-glucopyranoside, according to the Freudenberg method and crystallized from petroleum, m.p. 121–122°, $(\alpha)_D^{25} + 97.2^\circ$ (c. 1.0).

2,3 *Diacetyl-6-trityl-methyl- α ,D-glucopyranoside*. The acetals were esterified with acetic anhydride and then heated to 40° in 20 ml alcoholic 1% solution of HCl. The solution was neutralized with Ag_2CO_3 , trityl chloride added and the product after crystallization from ligroin yielded white crystals of the known 2,3 diacetyl-6-trityl-methyl- α ,D-glucopyranoside m.p. 160–162°, $(\alpha)_D^{25} + 76.2^\circ$ (c. 1.0 in $CHCl_3$).